Instructor:
Dr. Kristen Baker
Office: Caputo Hall 317
Phone: 717-871-7419
Email: Kristen.Baker@millersville.edu (I generally respond to emails promptly and on the same day. However, I do check my email less frequently after 8 pm and on the weekends, so please take this into consideration when contacting me with any questions you may have!)

Office Hours:
No appointment needed! Arrive with any questions you may have about the course material or come to listen to your classmate’s questions!
Monday 1:00-3:00 PM
Thursday 10:00 am -12:00 noon
Friday 11:00 am – 12:00 noon

If these times do not work for you, please use the calendly link below to schedule a 15-minute block of time to meet with me! You may schedule up to 30 minutes (2 blocks) a day. If you cannot find a time on calendly that works for you, please send me an email and we can work something out!
Calendly link: https://calendly.com/bakekr01/15-minute-meeting

I highly encourage you to visit my office hours for any help you may need in this course, whether it be to work together on a specific problem or topic, go over how to study for a quiz, or discuss your grade in the class. In my experience, those who attend my office hours typically saw an increase in their understanding of the course material, leading to a higher grade in the course. As mentioned earlier, I am also available to meet at additional times either one-on-one or in a small group, just use the calendly link to schedule! (https://calendly.com/bakekr01/15-minute-meeting)

Chemistry Peer Learning Hours:
No appointment needed! Simply come to work on chemistry with your peers and get help from experienced tutors. Starts August 29th
Location: Caputo 211
Tuesday 5-7 PM
Wednesday 12-2 PM, 5-7 PM
Thursday 12-2 PM, 5-7 PM

Required Class Meetings:
1. Lecture (CRN 4338)
   MWF, 10:00 AM – 10:50 AM
   Location: 211 Caputo Hall
2. Laboratory
   235 01A (CRN 4339) R 1:10 – 4:00 PM OR
   235 01B (CRN 4340) F 1:00 – 3:50 PM
   Location: 331 Caputo Hall

If you are unable to attend a class meeting, please email me asap.
Required Materials:

Please see Millersville University textbook store for more details.

D2L:
All important course information will be found on D2L. This includes lecture notes/videos, handouts, exams, quizzes, and answer keys. It is your responsibility to keep up with this material, including any information posted in an announcement.

Millersville Email:
Announcements may also be sent via email. Please check your email at least once a day.

Electronic Devices, Attendance, and In-Class Participation:
In order to succeed in this course, you need to attend class and be actively engaged in the lecture material by participating and taking meaningful notes. It is expected that while you are in lecture, your full attention is given to chemistry. Use of personal electronic devices will not be tolerated, except in cases where your device is used as an instructional tool (such as an iPad for notetaking/problem solving).

Inclusive Excellence Statement:
In this class, people of all different ethnicities, gender and gender identities, religions, ages, sexual orientations, disabilities, socioeconomic backgrounds, regions, and nationalities are encouraged to participate and share their perspectives and experiences that make each of us unique. Everyone will be expected to always treat each other with the utmost respect and consideration and without judgement. This classroom community will be a safe learning environment that encourages deep discussions based on everyone’s individual perspective on the concepts learned. Any suggestions you may have on how to improve the effectiveness of the course are always encouraged and appreciated. Please do not hesitate to talk to me about any concerns you have about your success in this class.

Course Description:
Organic chemistry is the study of the structure, bonding, and reactivity of molecules containing carbon. Throughout this course, you will gain a fundamental understanding of the reactivity of carbon-containing compounds, but I hope you also gain an appreciation for organic chemistry and its role in modern society, including pharmaceuticals, agriculture, plastics and more. Most importantly, you will develop your problem solving and critical thinking skills throughout this course. It is important to note that to succeed in this course (which I want you all to do!), you cannot just memorize the material. Instead, you must first understand the material and then learn how to apply this knowledge to how to approach problems, how to reason analytically, use logic, and how to communicate effectively. Throughout this journey we will do lots of practice problems, make mistakes (and learn from them), utilize potentially new learning strategies, and learn how to solve puzzles. Please do not hesitate to reach out with any problems or concerns you may have throughout the semester!
Course Goals:
1. Recognize and name simple organic molecules
2. Represent three-dimensional molecules with drawings in 2D, including analysis of energy states
3. Explain the attractive forces within and between molecules
4. Draw and analyze structural conformations, configurations, and isomers
5. Apply acid and base chemistry concepts to explain the reactivity of molecules
6. Describe the asymmetry of organic molecules, including non-standard chirality
7. Draw reasonable reaction mechanisms of a multitude of reactions including Frontier Molecular Orbital (FMO) notation
8. Predict plausible products based on starting materials and reaction conditions
9. Devise multi-step syntheses of simple organic molecules
10. Perform investigative and synthetic experiments using appropriate laboratory techniques

Course Content:
This course will cover the following chapters in your textbook (not specifically in this order):
Ch 1: Relevant General Chemistry
Ch 2: Functional Groups & Resonance
Ch 3: Acid-Base Reactions
Ch 4: Alkane Naming & Conformations
Ch 5: Stereoisomer Concepts
Ch 6: Reaction Mechanism Basics
Ch 17: Aromaticity & Acid-Base Reactions
Ch 19: Aldehydes, Ketones, & Derivatives
Ch 20: Carboxylic Acid Derivatives
Ch 21: Enols & Enolates
Ch 22: Amines

Grades:
Final grades will be based on the number of points you have earned collectively on homework/lecture problems, exams, quizzes, and laboratory work. You will have the opportunity to drop your three lowest quiz scores.

Homework/Lecture Problems 10%
Quizzes 18%
Four exams 40% (10% each)
Comprehensive final exam 12%
Laboratory* 20%
*A student cannot pass the course if they do not pass the laboratory portion.

Final grades will be based on the scale below. The minimum requirements for a particular grade may be lowered at my discretion.

<table>
<thead>
<tr>
<th>Grade</th>
<th>%</th>
</tr>
</thead>
<tbody>
<tr>
<td>A</td>
<td>&gt;93</td>
</tr>
<tr>
<td>A-</td>
<td>&gt;90</td>
</tr>
<tr>
<td>B+</td>
<td>&gt;87</td>
</tr>
<tr>
<td>B</td>
<td>&gt;83</td>
</tr>
<tr>
<td>C+</td>
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<td>D</td>
<td>&gt;63</td>
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<tr>
<td>D-</td>
<td>&gt;60</td>
</tr>
<tr>
<td>F</td>
<td>&lt;59</td>
</tr>
</tbody>
</table>

University Approved Class Attendance Policy:
1. **Students are expected to attend all classes.** It is the student's responsibility to complete all course requirements even if a class is missed. If a student misses class for an officially excused
reason, then he/she is entitled to make up the missed work but only at the convenience of the faculty member. Responsibility for materials presented in, assignments made for, and tests/quizzes given in regularly scheduled classes lies solely with the student.

2. The University policy is that faculty will excuse absences for the following reasons:
   - personal illness
   - death or critical illness in the family
   - participation in a university-sponsored activity
   - jury duty
   - military duties
   - religious holidays

3. Faculty judge the validity of student absences from class within the University's approved guidelines and may require documentation for excused absences. Faculty will evaluate any reason, other than those listed above, for a student missing class and determine whether the absence is justified. In these circumstances, a student may make up missed work at the discretion of the instructor.

4. In the case of foreseeable absences, students are encouraged to notify the faculty member in advance. A student who will miss class due to participation in an official University activity must notify the instructor well in advance of the activity to assure that the absence is excused.

Course Structure:
All material related to this course will be found on the D2L website. Before each lecture, I will post a 20-30 minute lecture video going over the concepts for that day’s lecture. You will be expected to watch the video before attending lecture.

During each lecture, I will do a brief review of the key points from the video. We will then split into small groups to work through the homework problems. Everyone is expected to participate in working on the problems. Throughout this time, I will be available to answer any questions. If you would like further clarification on any concepts or problems, please do not hesitate to ask. It is very likely others in the class have the same question.

Homework/Lecture Problems:
As detailed above, most of the time in our lecture meetings will be spent working through your homework problems either as a class or in a small group, which is an important way of learning the material for this course. Participating in these practice problems will be crucial in your ability to succeed in this class. As such, I expect everyone to attend each lecture. Each day’s homework problems are due at the beginning of the next class. I do not expect you to always know the correct answer; instead, I am looking for evidence that you have spent time trying to understand the material. Having the skills to think out loud and reason through a problem will benefit you throughout your career.

Quizzes:
Daily quizzes will be given throughout the semester during the first 10 minutes of lecture. Two points of each quiz will be a “reading quiz” based on the lecture material for that day. The other 8 points will be from questions on the previous day’s lecture. Your lowest three quiz grades will be dropped. Because I drop your three lowest grades, no makeup quizzes will be given. Tokens earned throughout the semester may be exchanged for a quiz retake. Any unused tokens will grant
you 1 extra point on the final exam. To earn your first token, send me an email by September 1st with the subject “token”. In this email, let me know anything in the syllabus you are confused about and send me your favorite cheesy joke!

Exams:
There will be four exams given throughout the course of the semester. You will be expected to show your work on the exams. Exams will begin promptly at the beginning of class; if you arrive late, you will not be given extra time. Absence from an exam will only be excused if you provide adequate documentation covering the excuse.

How I grade:
All points on exams and quizzes are earned rather than points being taken off for mistakes. Please be sure to read all instructions carefully and to provide a complete answer which may include pictures or diagrams. Your goal should be to show me that you fully understand the material that you are being asked about.

Exam Coverage and Tentative Dates:
Exam #1: Chapters 1-3  Friday, 9/15
Exam #2: Chapters 4-6  Friday, 10/06
Exam #3: Chapters 17, 19-20  Friday, 11/03
Exam #4: Chapters 21-22  Friday, 12/01

Cumulative Final Exam Date: 8AM – 10 AM on Thursday, December 7th, 2023

Study Recommendations:
A good rule of thumb for any college course is allotting 3-4 hours of studying per each lecture hour. After each lecture, you should review and study your class notes. Once you feel prepared, you should then attempt the problems handed out during class, either on your own or in small groups. I highly recommend forming study groups! The best way to study for this course is to actively solve problems. This does not have to all be done in one sitting, rather spend small periods of time on solving problems each day. During these study periods, ask yourself if you are using your study time and methods effectively. Organic chemistry requires your attention and practice every single day. It is to your advantage to keep up with your work as each subsequent chapter builds on the previous chapter. If you find yourself getting behind, please feel free to stop by my office so we can make a study schedule! In addition to studying by yourself or in a small group, I highly encourage you to attend my office hours.

Organic chemistry requires a lot of actively practicing problems….you cannot just memorize! Subsequent chapters build on material you learned previously, so you must know the early material like the back of your hand if you want to be able to tackle the harder material.

Potentially Useful Resources:
- Evans’ pKa Table: http://goo.gl/f6D15q
- Structural Database of Organic Compounds (SDBS), a database of compounds and their physical and spectroscopic properties: http://goo.gl/5AoXa
Important Dates:
- **August 28th, 2023** – last day to DROP/ADD classes online; last day to request PASS/FAIL or AUDIT
- **October 27th, 2023** – last day to DROP classes (with a ‘W’ grade)

Academic Conduct Code:
Academic dishonesty will not be tolerated. Not only is cheating unethical and disrespectful to your faculty and fellow students, it is also self-destructive to your own academic integrity. All students are expected to maintain high standards of academic integrity; you are responsible for understanding and abiding by the Millersville University Academic Honesty Policy outlined below. *If you break this policy, you will be assigned a failing grade and may be prosecuted by the Academic Review Board.* While you may work together on some assignments, all work turned in must be your own work and answers must be written in your own words. Forms of academic dishonesty include (but are not limited to) the use of cheat sheets during exams, copying answers from other students, plagiarizing material, knowingly allowing others to represent your work as their own, and use of online databases that provide “expert” answers to posted questions such as Chegg, Slader, CourseHero, and Koofers. Additionally, I own the copyright to all course materials, which may not be duplicated in any form other than for your own learning. This includes uploading these materials to any online sites that will provide “expert” answers or giving/receiving old course materials from other students.

Millersville University’s Academic Honesty Policy: Students of the University are expected to be honest and forthright in their academic endeavors. Actions that violate the Academic Honesty Policy include:
- Plagiarism (using someone else’s words, ideas, or data)
- Fabrication (falsifying results in research or other findings)
- Cheating (the act or attempted act of deception by which an individual tries to misrepresent that he/she has mastered subject matter in an academic project or the attempt to gain advantage by the use of illegal or illegitimate means)
- Academic Misconduct: violation of university policies by tampering with grades or participating in the distribution of any part of a test before its administration.

For more information:
[https://www.millersville.edu/learningservices/](https://www.millersville.edu/learningservices/)

Title IX Statement:
Title IX Reporting Requirements and the Faculty member: Millersville University is committed to maintaining a safe education environment for all students. In compliance with Title IX of the
Education Amendments of 1972 and guidance from the Office for Civil Rights, the University requires faculty members to report incidents of sexual violence shared by students to the University’s Title IX Coordinator. The only exceptions to the faculty member’s reporting obligation are when incidents of sexual violence are communicated by a student during a classroom discussion, in a writing assignment for a class, or as part of a University-approved research project. Faculty members are obligated to report sexual violence or any other abuse of a student who was, or is, a child (under 18 years of age) when the abuse allegedly occurred to the person. Information about Title IX, resources and reporting can be found at: What is Title IX | Millersville University

Land Acknowledgement:
We would like to recognize the Native peoples of the lower Susquehanna River basin, those known and those unknown to us, who have stewarded the land, upon which Millersville University sits, for thousands of years. We acknowledge that the land on which we gather, study, and work is the ancestral land of the Conestogas, Susquehannocks, Shawnee, and others. One group, the Shenks Ferry people, had a village adjacent to the campus. We pay our respects to the traditional occupants and caretakers of this land.

Counseling Reminder:
Students sometimes face mental health or drug/alcohol challenges in their academic careers that interfere with their academic performance goals. Millersville is a caring community and resources are available to assist students who are dealing with problems.

- Counseling Center (717-871-7821)
- Health Services (717-871-5250)
- Center for Health Education and Promotion (717-871-4141)
- Campus Ministries
- Learning Services (717-871-5554)

Americans with Disabilities Act:
Millersville University is committed to equality of opportunity and freedom from discrimination for all students, employees, applicants for admission or employment, and all participants in public University-sponsored activities. In keeping with this commitment, and in accordance with the Americans with Disabilities Act (ADA) the University will make every effort to provide equality of opportunity and freedom from discrimination for all members of the University community and visitors to the University, regardless of any disability an individual may have. Accordingly, the University has taken positive steps to make University facilities accessible to individuals with disabilities to participate in University programs. The University administration and management are obligated to report any allegation of discrimination to the appropriate office as defined in this policy.

Questions?? Concerns?? Stop by my office hours!
Monday 1:00-3:00 PM
Thursday 10:00 am -12:00 noon
Friday 11:00 am – 12:00 noon
## Tentative Course Schedule:

<table>
<thead>
<tr>
<th>Week/Dates</th>
<th>Topics Covered</th>
<th>Reading</th>
<th>Notes</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Week #1</strong> 8/21–8/25</td>
<td>Introduction Lewis Structures, Valence Bond Theory Molecular Orbital Theory</td>
<td>1.1-1.14</td>
<td>Token Opportunity: Office Hours Meet and Greet</td>
</tr>
<tr>
<td><strong>Week #2</strong> 8/28–9/01</td>
<td>Line angle notation, functional groups Chemical arrows, resonance Acids/bases</td>
<td>2.1-2.13 3.1-3.4</td>
<td></td>
</tr>
<tr>
<td><strong>Week #3</strong> 9/06–9/08</td>
<td>Acids/bases</td>
<td>3.3-3.9</td>
<td>No Class Monday- Labor Day!</td>
</tr>
<tr>
<td><strong>Week #4</strong> 9/11–9/15</td>
<td>Lewis acids/bases</td>
<td>3.10</td>
<td>Exam #1 Friday, 9/15</td>
</tr>
<tr>
<td><strong>Week #5</strong> 9/18–9/22</td>
<td>Nomenclature/simple conformations Substituted conformations Cyclohexane</td>
<td>4.1-4.15</td>
<td></td>
</tr>
<tr>
<td><strong>Week #6</strong> 9/25–9/29</td>
<td>CIP, chirality Stereoisomers, configuration Nucleophiles and Electrophiles</td>
<td>5.1-5.11 6.6-6.10</td>
<td>Token Opportunity: Midsemester Survey</td>
</tr>
<tr>
<td><strong>Week #7</strong> 10/02–10/06</td>
<td>Carbocations, Reversible/Reversible Reaction Arrows</td>
<td>6.11-6.12</td>
<td>Exam #2 Friday, 10/06</td>
</tr>
<tr>
<td><strong>Week #8</strong> 10/11–10/13</td>
<td>Aromaticity Aldehydes/Ketones: Nuc addition</td>
<td>17.1-17.5 19.1-19.4</td>
<td>No Class Monday- Fall Recess!</td>
</tr>
<tr>
<td><strong>Week #9</strong> 10/16–10/20</td>
<td>Aldehydes/Ketones: different Nucs</td>
<td>19.5-19.11</td>
<td></td>
</tr>
<tr>
<td><strong>Week #10</strong> 10/23–10/27</td>
<td>Carboxylic Acids Acid Chlorides Acid Anhydrides, Esters</td>
<td>20.1-20.10</td>
<td></td>
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<tr>
<td><strong>Week #11</strong> 10/30–11/03</td>
<td>Esters, amides, nitriles</td>
<td>20.11-20.13</td>
<td>Exam #3 Friday, 11/3</td>
</tr>
<tr>
<td><strong>Week #12</strong> 11/6–11/10</td>
<td>Enols/Enolates Aldol Claisen</td>
<td>21.1-21.4</td>
<td></td>
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<tr>
<td><strong>Week #13</strong> 11/13–11/17</td>
<td>Alkylation Conjugate Addition Synthesis</td>
<td>21.5-21.7</td>
<td></td>
</tr>
<tr>
<td><strong>Week #14</strong> 11/20–11/24</td>
<td>Amine overview</td>
<td>22.1-22.3</td>
<td>No Class Wednesday or Friday- Thanksgiving Recess!</td>
</tr>
<tr>
<td><strong>Week #15</strong> 11/27–12/01</td>
<td>Amine reactions</td>
<td>22.4-22.6</td>
<td>Exam #4 Friday, 12/1</td>
</tr>
<tr>
<td><strong>Week #16</strong> 12/4–12/8</td>
<td>Final exam review</td>
<td></td>
<td>Finals begin Tuesday!</td>
</tr>
</tbody>
</table>

**Final Exam:** Thursday, 12/7 from 8-10 AM

**Disclaimer:** This syllabus and calendar may change over the course of the semester. All changes will be communicated in class and through a Canvas announcement or email.
Laboratory for Chemistry 235
Fall 2023

Laboratory Course Goals:
1. Demonstrate proper laboratory safety and waste disposal when working in the lab
2. Keep a neat and organized record of laboratory data in a notebook.
3. Set up apparatus for experimental techniques: reactions, distillations, filtrations, etc.
4. Purify organic products by recrystallization (solids) and distillation (liquids).
5. Characterize organic products by physical, chemical, and spectroscopic properties.

Laboratory Schedule:

<table>
<thead>
<tr>
<th>WEEK</th>
<th>ACTIVITY</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Safety, Check-in, and Core Concepts</td>
</tr>
<tr>
<td>2</td>
<td>Functional Groups and IR Spectroscopy</td>
</tr>
<tr>
<td>3</td>
<td>Recrystallization</td>
</tr>
<tr>
<td>4</td>
<td>Limonene Extraction via Steam Distillation</td>
</tr>
<tr>
<td>5</td>
<td>Acid-Base Reactions and Acid-Base Extraction</td>
</tr>
<tr>
<td>6</td>
<td>Conformational Analysis, Chirality, and Stereochemistry</td>
</tr>
<tr>
<td>7</td>
<td>Thin-Layer-Chromatography (TLC)</td>
</tr>
<tr>
<td>8</td>
<td>Proton and Carbon NMR</td>
</tr>
<tr>
<td>9</td>
<td>Oxone Oxidation of an aldehyde into a carboxylic acid</td>
</tr>
<tr>
<td>10</td>
<td>Acid cat. Cyclic acetal formation from benzaldehyde and pentaerythritol</td>
</tr>
<tr>
<td>11</td>
<td>Imine formation from an aldehyde (Multi-Step Synthesis- Step 1)</td>
</tr>
<tr>
<td>12</td>
<td>Imine formation from an aldehyde (Multi-Step Synthesis- Steps 2 and 3)</td>
</tr>
<tr>
<td>13</td>
<td>More Proton and Carbon NMR</td>
</tr>
<tr>
<td>14</td>
<td>Thanksgiving Break</td>
</tr>
<tr>
<td>15</td>
<td>Clean-up and Check-out</td>
</tr>
</tbody>
</table>

Chemistry 235 Laboratory Attendance Statement
Make-up labs are not an easy option (due primarily to scheduling and space limitations). If you are planning to miss a lab due to an official Millersville University event, or due to a foreseeable life event absence, or if you miss a lab due to an emergency, please email, as soon as possible, to begin discussing options; failure to begin discussing options within a timely manner will lead to this course policy: missing the first lab equals a zero for the lab activity, missing two labs equals a zero for the second lab activity, missing three or more labs equals failure of the entire course.

Academic Support Services:
Please see the Office of Learning Services in Lyle Hall as soon as possible if you have special learning needs for this class. If you have a condition that may affect your ability to perform laboratory exercises, to exit lab safely from the premises in an emergency, or which may cause an emergency during class, or lab, please discuss this in confidence with your instructor and someone at the Office of Student Support Services. Appropriate accommodations may then be provided.
http://www.millersville.edu/learningservices/